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Structure of Dichlorobis(triphenylphosphine oxide)manganese(II)

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Abstract. $[MnCl_2 \{PO(C_6H_5)_3\}_2], M_r = 682.43,$ orthorhombic, Fdd2, a = 20.990 (3), b = 33.028 (6), c = 9.756 (8) Å, V = 6763 (6) Å³, Z = 8, $D_m =$ 1.36 (1), $D_x = 1.340$ (1) g cm⁻³, λ (Mo Ka) = 0.71069 Å, $\mu = 6.16$ cm⁻¹, F(000) = 2808, T = ca293 K, R = 0.046 for 1137 observed reflections. The Mn atom lies on a twofold axis and is tetrahedrally coordinated. The structure displays no unusual features.

Experimental. The title complex was synthesized by Goodgame & Cotton (1961) and characterized by Davolos (1978). D_m by flotation (CH₃OH + CHBr₃), pale-yellow plates, $0.4 \times 0.3 \times 0.2$ mm, CAD-4 diffractometer, unit-cell dimensions from 25 reflections ignored, $(11 < \theta < 18^{\circ}),$ absorption $\theta < 23^{\circ}$, $0 \le h \le 23$, $0 \le k \le 36$, $0 \le l \le 10$, three standard reflections showed no appreciable intensity variation, 1222 independent reflections, 1137 observed reflections $[I > 3\sigma(I)]$, Lorentz and polarization corrections. Structure solved by MULTAN78 (Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978) and difference Fourier maps. Refinement by least squares using

Table	1. Ato	mic coordina	tes (×10 ⁴ e	xcepti	ng z coore	din-
ates of	f C:	$\times 10^3$) and	equivalent	isotr	opic ther	mal
param	eters	(Hamilton,	1959)	with	e.s.d.'s	in
		par	entheses			

	x	у	Z	$B_{eq}(\text{\AA}^2)$
Mn	0	0	0	2.83 (5)
Cl	750(1)	344 (1)	-1251 (4)	5.5(1)
Р	870(1)	-709 (1)	1862 (3)	3.12(7)
0	410 (3)	-397 (2)	1383 (7)	4.1 (2)
C(1)	1685 (4)	-552 (3)	168 (1)	3.3 (3)
C(2)	1931 (5)	-512(3)	38 (1)	4.8 (4)
C(3)	2570 (5)	-386 (3)	23 (1)	5.4 (4)
C(4)	2944 (5)	-305 (3)	136 (1)	5.0 (4)
C(5)	2692 (5)	-340 (3)	266 (2)	5.6 (5)
C(6)	2055 (5)	-471 (3)	285(1)	5.0 (4)
C(7)	731 (4)	-808 (3)	366 (1)	3.6 (3)
C(8)	391 (8)	-519 (4)	438 (1)	7.6 (6)
C(9)	300 (10)	-559 (5)	578 (2)	10.2 (8)
C(10)	530 (8)	-908 (5)	644 (1)	7.9 (6)
C(11)	859 (8)	-1197 (4)	573 (2)	6.9 (6)
C(12)	965 (6)	-1149 (3)	431 (1)	5.5 (4)
C(13)	792 (4)	-1169 (3)	91 (1)	3.6 (3)
C(14)	1315 (5)	-1440 (3)	76 (1)	5.1 (4)
C(15)	1246 (6)	-1787 (3)	-7 (2)	6.1 (5)
C(16)	670 (7)	-1865 (3)	-76(1)	6.3 (5)
C(17)	146 (7)	-1590 (4)	-60(1)	6.5 (5)
C(18)	208 (5)	-1238 (3)	24 (1)	4.5 (4)

SHELX76 (Sheldrick, 1976), $\sum w(|F_o| - |F_c|)^2$ minimized, atomic coordinates, anisotropic thermal parameters and scale factor refined, isotropic thermal parameters for H atoms, final R 0.046, unit weights, maximum Δ/σ in final refinement cycle 0.04, maximum and minimum height in final difference Fourier map 0.3 and $-0.5 \text{ e} \text{ Å}^{-3}$. Scattering factors from *International Tables for X-ray Crystallography* (1974). Calculations performed on the VAX 11/780 computer at the Instituto de Fisica e Quimica de São Carlos, USP, Brazil. Positional and equivalent isotropic thermal parameters are listed in Table 1.* The atom-numbering scheme, bond lengths and angles are shown in Fig. 1. A view of the molecule is shown in Fig. 2 (Johnson, 1965).

^{*} Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42372 (7 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. Numbering scheme and bond lengths (Å) and angles (°) with e.s.d.'s in parentheses.

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Fig. 2. View of the molecule.

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Structure of a Bicyclic Sulfur–Nitrogen–Carbon Heterocyclic Molecule

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Abstract. 7-Phenyl- $1\lambda^4, 3\lambda^4\delta^2, 5\lambda^4$ -trithia-2,4,6,8,9-pentaazabicyclo[3.3.1]nonane, $C_7H_5N_5S_3$, $M_r = 255\cdot3$, monoclinic, $P2_1/n$, a = 5.958 (1), b = 22.954 (2), c =7.428 (1) Å, $\beta = 106\cdot25$ (1)°, $V = 975\cdot2$ (4) Å³, Z = 4, $D_x = 1.74$ g cm⁻³, λ (Mo Ka) = 0.71073 Å, $\mu =$ 7.00 cm⁻¹, F(000) = 520, T = 293 K, R = 0.030 for 1363 unique observed reflections. The planar SNCNS and SNSNS components of the bicyclic molecule make dihedral angles of 137.1 and 118.3° with the SNS bridging unit. The S–N bonds connecting the NSN fragment to the CN₃S₂ ring are much longer (1.728 Å) than those in the remainder of the molecule (1.546– 1.630 Å).

Experimental. Compound prepared by reaction of $N_1N_2N_3N_3Cl_3$. Crystals obtained from acetonitrile solutions. Yellow

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needle data crystal $0.60 \times 0.16 \times 0.14$ mm mounted on a glass fiber. Density not measured. Intensities measured with an Enraf-Nonius CAD-4 diffractometer using variable-speed ω -2 θ scans. Unit cell determined from least-squares analysis of angle data for 25 reflections with $20 < 2\theta < 28^{\circ}$. Analytical absorption correction based on crystal shape varied from 0.95 to 1.00. Data collected to $\sin\theta/\lambda$ of 0.59 Å⁻¹, $-7 \le h \le 7$, $0 \le k \le 27, 0 \le l \le 8$. Three standard reflections (252, $1,12,0, \bar{3}73$) varied $\pm 4.1\%$ over 16.9 h of data collection; anisotropic-drift correction applied. 1839 reflections measured, 1708 unique $(R_{int} = 0.01)$, 345 reflections with $I < 3\sigma(I)$ considered unobserved. Solved by direct methods using MULTAN11/82 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982) and Fourier methods. Full-matrix least squares minimized $\sum w \Delta F^2$. H atoms refined isotropically, other

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